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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.002 Å R factor = 0.037 wR factor = 0.088 Data-to-parameter ratio = 13.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_{34}H_{28}Cl_2O_2$, was synthesized by the selfcoupling reaction of 2-(4-chlorobenzyl)-1-tetralone induced by low-valent titanium reagent (TiCl₄/Zn) in tetrahydrofuran. The X-ray analysis reveals that the two cyclohexenone rings adopt different conformations, *viz*. envelope and distorted half-chair.

2,2'-[1,2-Bis(4-chlorophenyl)ethane-

1,2-diyl]bis(1-tetralone)

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Comment

In the early seventies, Tyreik & Wolochowicz (1973), Mukaiyama *et al.* (1973) and McMurry & Fleming (1974) established that low-valent titanium can abstract oxygen from ketones or aldehydes leading to formation of olefins. An increasing interest in coupling reactions induced by low-valent titanium reagents has been observed and a large number of functional groups can be reduced (Shi *et al.*, 1993, 1997, 1998). We report here the crystal structure of the title compound, (I), synthesized by the self-coupling reaction of 2-(4-chlorobenzyl)-1-tetralone induced by low-valent titanium reagent (TiCl₄/Zn).



The molecular structure of (I) is shown in Fig. 1 and selected bond lengths and angles are listed in Table 1. The C7-C24 linkage resulting from dimerization reaction of 2-(4chlorobenzyl)-1-tetralone has a length of 1.549 (2) Å. The bond angles around C7 and C24 deviate from the ideal value for a sp^3 carbon. One of the cyclohexenone rings (C8–C11/ C16/C17) adopts an envelope conformation with C9 deviating by 0.652 (2) Å from the plane determined by the other five atoms. The other cyclohexenone ring (C25-C28/C33/C34) adopts a distorted half-chair conformation, with atoms C25 and C26 deviating from the C27/C28/C33/C34 plane by -0.186(2) and 0.576(2) Å, respectively. In addition, the X-ray analysis reveals that the configuration of (I) is threo and its conformation is gauche. Carbonyl atoms O1 and O2 are involved in weak intramolecular C-H···O interactions (Table 2) and the molecular packing in the crystal (Fig. 2) is stabilized by C-H··· π interactions involving C30-H30 and phenyl ring (C18–C23) at (x, 1 + y, z).

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 $R_{\rm int} = 0.015$

 $\theta_{\rm max} = 25.0^{\circ}$

 $h = 0 \rightarrow 12$ $k = -12 \rightarrow 12$

 $l = -15 \rightarrow 14$

3 standard reflections

every 97 reflections

intensity decay: 4.3%

 $w = 1/[\sigma^2(F_o^2) + (0.0415P)^2]$

Extinction correction: *SHELXTL* Extinction coefficient: 0.0082 (11)

+ 0.0134*P*] where $P = (F_o^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.20 \text{ e } \text{\AA}^{-3}$



Figure 1

The molecular structure of (I), showing 50% probability of displacement ellipsoids and the atom-numbering scheme.



Figure 2

The molecular packing of (I), viewed down the c axis.

Experimental

The title compound, (I), was prepared by the self-coupling reaction of 2-(4-chlorobenzyl)-1-tetralone induced by low-valent titanium reagent ($TiCl_4/Zn$); m.p. 491–493 K. Single crystals were obtained by slow evaporation of an ethanol solution.

Crystal data

$C_{34}H_{28}Cl_2O_2$	Z = 2
$M_r = 539.46$	$D_x = 1.339 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Mo $K\alpha$ radiation
a = 10.411 (2) Å	Cell parameters from 31
b = 10.975(1) Å	reflections
c = 13.296 (2) Å	$\theta = 4.4 - 15.0^{\circ}$
$\alpha = 103.54 \ (1)^{\circ}$	$\mu = 0.27 \text{ mm}^{-1}$
$\beta = 112.97 \ (1)^{\circ}$	T = 294 (2) K
$\gamma = 93.89 \ (1)^{\circ}$	Block, colorless
$V = 1338.2 (4) \text{ \AA}^3$	$0.52 \times 0.48 \times 0.48$ mm

Data collection

Siemens P4 diffractometer ω scans Absorption correction: ψ scan (XSCANS; Siemens, 1994) $T_{min} = 0.865, T_{max} = 0.877$ 5121 measured reflections 4671 independent reflections 3392 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.088$ S = 1.054671 reflections 344 parameters H-atom parameters constrained

Table 1

Selected geometric parameters (Å, °).

Cl1-C1	1.7420 (18)	C7-C24	1.549 (2)
Cl2-C18	1.7409 (19)	C7-C8	1.555 (2)
O1-C17	1.214 (2)	C21-C24	1.526 (2)
O2-C34	1.217 (2)	C24-C25	1.546 (2)
C4-C7	1.520 (2)		
C4-C7-C24	111.85 (14)	C21-C24-C25	110.19 (14)
C4-C7-C8	108.79 (14)	C21-C24-C7	111.58 (14)
C24-C7-C8	115.78 (14)	C25-C24-C7	114.44 (14)
O1-C17-C16	120.68 (17)	O2-C34-C33	121.16 (16)
O1-C17-C8	121.60 (17)	O2-C34-C25	121.86 (16)
C16-C17-C8	117.68 (15)	C33-C34-C25	116.91 (15)
C4-C7-C24-C21	-59.28 (18)	C4-C7-C24-C25	174.74 (14)
C8-C7-C24-C21	175.33 (13)	C8-C7-C24-C25	49.4 (2)

Table 2Hydrogen-bonding geometry (Å, $^{\circ}$).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
C7-H7···O1	0.98	2.37	2.820 (2)	107
C20-H20···O2	0.93	2.55	3.183 (3)	125
$C24 - H24 \cdots O2$	0.98	2.37	2.797 (2)	106
$C30-H30\cdots CgP^{i}$	0.93	2.72	3.626 (3)	166

Symmetry code: (i) x, 1 + y, z. CgP is the centroid of the phenyl ring C18–C23.

H atoms were fixed geometrically and allowed to ride on their parent C atoms, with C–H distances fixed in the range 0.93–0.98 Å and $U_{iso}(H)$ values set equal to $1.2U_{eq}(C)$.

Data collection: *XSCANS* (Siemens, 1994); cell refinement: *XSCANS*; data reduction: *SHELXTL* (Sheldrick, 1997); program(s) used to solve structure: *SHELXTL*; program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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